



Thomas Haney Secondary School

Science 10

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LG 3: Energy in Chemical Processes

BIG IDEA: How is energy involved in chemical processes?

| Learning Standards: | |
|---|---|
| <p>Fundamental Knowledge <i>(what I need to know)</i></p> <ul style="list-style-type: none"> ☆ <i>Energy is involved in chemical processes</i> ☆ <i>Chemical reactions occur in predictable ways (single replacement, double replacement, etc.).</i> ☆ <i>Chemical reactions result from the rearrangements of atoms and their bonds.</i> | <p>Curricular Competencies <i>(What I need to do)</i></p> <ul style="list-style-type: none"> • I can seek and analyze patterns, trends, and connections in data, including describing relationships between variables and identifying inconsistencies • I can formulate hypothesis and probable outcomes. • I can use knowledge of scientific concepts to draw conclusions that are consistent with the evidence. |

| Assessment of Learning Standards: | | |
|---|--|---|
| Have an interview to show evidence of the Learning Standards . | | |
| | ADVANCED (B) | MASTERY (A) |
| <p style="text-align: center;">ESSENTIALS (C/C+)</p> <p>I CAN:</p> <ul style="list-style-type: none"> ☆ Recognize the different reaction types (single replacement, double replacement, combustion (oxidation), acid/base neutralization, synthesis, and decomposition. ☆ Explain how atoms are rearranged in each of the reaction types. ☆ List the different reaction types and provide an example of each. | <p>I CAN:</p> <ul style="list-style-type: none"> ➤ Predict the products of a reaction give the reactants AND List the different reaction types and provide an example of each. ➤ Explain how conservation of mass and energy changes are relate (ie. Endothermic vs Exothermic). | <p>I CAN:</p> <ul style="list-style-type: none"> ✓ Balance the examples of the different reaction types and explain how they prove that mass is conserved. |

Reflection:

After finishing my learning activities what do I understand? How have I answered the BIG Question?

OPTION 1

- ☆ **Read** BC Science 10 pgs. 256 - 267 to learn about Types of Chemical Reactions and complete Pg. 105-110 from the Section 6.1 Worksheets.
- ☆ **Read** BC Science 10 pgs. 202 - 211 to learn about Balancing Chemical Reactions and complete Pg. 77-79 from the Section 4.3 Worksheets.
- ☆ Go to the following website: <https://www.youtube.com/watch?v=L-G7pLufXAo> and **complete** the Bozeman Science worksheet
- ☆ **Complete** the “Chemical Reaction Types” lab by completing the attached worksheet.

OPTION 2

- ☆ **Read** BC Science 10 pgs. 256 - 267 to learn about Types of Chemical Reactions and do questions on p. 271.
- ☆ **Read** BC Science 10 pgs. 202 - 211 to learn about Balancing Chemical Reactions and do questions on p. 215.
- ☆ Go to the following website: <https://www.youtube.com/watch?v=L-G7pLufXAo> and **complete** the Bozeman Science worksheet
- ☆ **Complete** the “Chemical Reaction Types” lab by completing the attached worksheet.